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Attorney Docket No.: DN 99-009

OCT 2 3 2007

Appl. No. 10/526,342 AMENDMENT dated 10/23/07

In the Claims:

Please cancel claims 3 and 11.

- 1. (Currently Amended) A process for producing precipitated calcium carbonate, comprising the steps of:
 - (a) providing calcium hydroxide having at least about 92 weight percent solids; and
- (b) carbonating the calcium hydroxide with carbon dioxide gas and simultaneously comminuting for a time sufficient to produce a calcium carbonate having at least about a 90 weight percent conversion to calcium carbonate and having a solids concentration of at least about 92 weight percent.
- 2. (Canceled)
- 3. (Canceled)
- 4. (Previously Presented) The process for producing precipitated calcium carbonate according to claim 1, wherein the calcium hydroxide provided in step (a) is produced by the steps comprising:
 - i) mixing calcium oxide and water in amounts sufficient to react to form calcium hydroxide substantially free of water; and
 - maintaining the mixture at an elevated temperature for a time sufficient to hydrate the calcium oxide to form calcium hydroxide having at least about 92 weight percent solids.

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percent conversion to calcium hydroxide.

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5. (Original) The process according to claim 4, wherein the step of maintaining the mixture at an elevated temperature is performed at a temperature of up to about 600 degrees Fahrenheit for a time sufficient to hydrate the calcium oxide to form calcium hydroxide having at least about a 95 weight

6. (Original) The process according to claim 5, wherein the step of maintaining the mixture at an elevated temperature is performed for a time sufficient to hydrate the calcium oxide to form calcium hydroxide having at least about a 98 weight percent conversion to calcium hydroxide.

7. (Previously Presented) The process according to claim 1, wherein the steps of carbonating and simultaneously comminuting are performed until at least a 95 weight percent conversion to calcium carbonate is achieved.

8. (Previously Presented) The process according to claim 1, wherein the steps of carbonating and simultaneously comminuting are performed until at least a 97 weight percent conversion to calcium carbonate is achieved.

- 9. (Currently Amended) A process for producing precipitated calcium carbonate, comprising the steps of:
 - (a) providing calcium hydroxide having at least about 92 weight percent solids;
- (b) carbonating the calcium hydroxide with carbon dioxide gas for a time sufficient to at least partially convert the calcium hydroxide to calcium carbonate;

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- (c) comminuting the at least partially converted calcium hydroxide; and
- (d) repeating alternating steps of carbonating and comminuting for a time sufficient to substantially convert the calcium hydroxide to calcium carbonate having at least about a 90 weight percent conversion to calcium carbonate and having a solids concentration of at least about 92 weight percent.
- 10. (Canceled)
- 11. (Canceled)
- 12. (Previously Presented) The process for producing precipitated calcium carbonate according to claim 9, wherein the calcium hydroxide provided in step (a) is produced by the steps comprising:
 - mixing calcium oxide and water in amounts sufficient to react to form calcium
 hydroxide substantially free of water; and
 - maintaining the mixture at an elevated temperature for a time sufficient to hydrate the calcium oxide to form calcium hydroxide having at least about 92 weight percent solids.
- 13. (Original) The process according to claim 12, wherein the step of maintaining the mixture at an elevated temperature is performed at a temperature of up to about 600 degrees Fahrenheit for a time sufficient to hydrate the calcium oxide to form calcium hydroxide having at least about a 95 weight percent conversion to calcium hydroxide.

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14. (Original) The process according to claim 13, wherein the step of maintaining the mixture at an elevated temperature is performed for a time sufficient to hydrate the calcium oxide to form calcium hydroxide having at least about a 98 weight percent conversion to calcium hydroxide.

15. (Original) The process according to claim 9, wherein the steps of carbonating and comminuting are performed until at least a 95 weight percent conversion to calcium carbonate is achieved.

16. (Original) The process according to claim 15, wherein the steps of carbonating and comminuting are performed until at least a 97 weight percent conversion to calcium carbonate is achieved.

- 17. (Canceled)
- 18. (Canceled)
- 19. (Canceled)
- 20. (Canceled)